



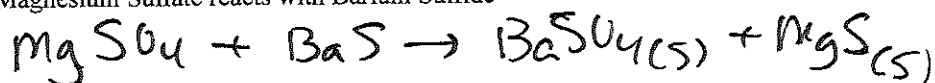
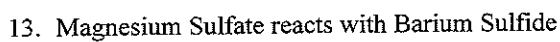
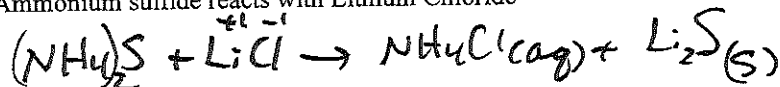
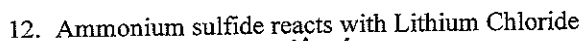
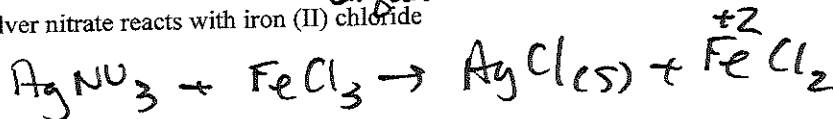
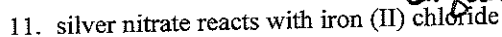
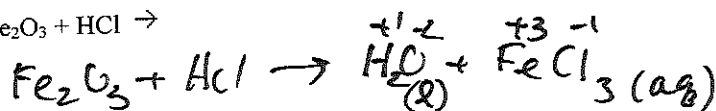
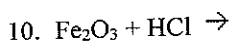
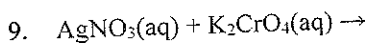
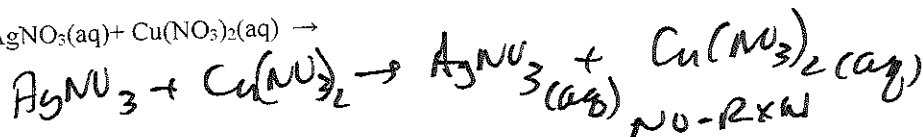
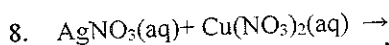
STUDENT PRACTICE FOR PROFICIENCY

Chemistry
Solubility Reactions #1Molecular Equation: $AB + CD \rightarrow CB_{(aq)} + AD_{(s)}$ Insoluble = precipitate
Soluble = aqueous***Need at least 1
insoluble product or
water

Quick Check

 Na^+X K^+X NH_4^+X XNO_3^- Always
Soluble

- Determine potential products
- Indicate if the product will be soluble(s), insoluble (aq), or Liquid ($H_2O(l)$)
- If there are no "NEW" products indicate no reaction.



two Solids